VIIth International Conference on Raman Spectroscopy

Réclacteur - W.F. Murphy - Editor

100

1980

NORTH-HOLLAND PUBLISHING COMPANY

AMSTERDAM NEW YORK

SIZE EFFECTS ON THE RAMAN EXCITATION PROFILES OF MIXED-VALENCE COMPOUNDS WITH METAL-HALIDE CHAIN

G.C. Papavassiliou

Theoretical and Physical Chemistry Institute, National Hellenic Research Foundation, 48, Vassileos Constantinou Avenue, Athens 501/1, Greece

ABSTRACT - The Raman excitation profile of a metal - halide chain compound occurs near the gap frequency. The position of profiles of small particles depends strongly on the size of particles.

RÉSUMÉ - On montre que le profil d'excitation est situé au voisinage de la fréquence de gap. La position des profils de petites particules dépend de la taille de particules.

In our previous work /1/ - /3/ we showed that the position, the shape and the intensity of bands in the absorption spectra of mixed-valence compounds with a metal-halide chain depend on the size, the shape and the aggregation of the crystals. We also observed a small shift in the Raman excitation profiles (REP) when we changed the dispersed medium /3/. In this paper we describe the changes(position, shape, intensity) in the REP when one changes the size (the shape, and the aggregation) of crystals of mixed valence compounds with the chain-M(II)--X-M(IV)--(M=Pt or Pd, X=Cl, Br or I). We consider as an example the compound Pt(dapn) Pt (dapn) Br (ClO₄) (where dapn=1,2-diaminopropane) of which the absorption bands occur in the visible region /1/,/2/.

Pt(dapn)₂Pt(dapn)₂Br₂(ClO₄)₄ has been prepared as in /4/ and recrystallized twice from water containing small amount of HClO_{μ} . Large crystals (1x1x) have been obtained by slow cooling aqueous solutions of the compound. It is soluble in water, methanol, acetonitile, etc. but insoluble in carbon tetrachloride, carbon disulphide and ether. We made use of this property to prepare suspensions of the complexes in the last three liquids. The samples were prepared as follows.(A) 25mm of saturated solution of the compound in acetonitrile were injected slowly into 20cm carbon tetrachloride which was stirred vigorously, The suspension obtained was red in colour.(B) 25mm of the same saturated solution in acetonitrile were added at once to 2cm carbon tetrachloride without stirring; after 1min, 18cm carbon tetrachloride were added and the suspension was stirred. This suspension was violet in colour. Electron micrographs showed that the samples(A)and (B) consisted of small particles 150 and 900% in diameter respectively. We also prepared thin diposits of compound on quartz plates /1/-/3/. These deposits consist of small needles(1000-2000A) and are blue in colour. After rubbing the deposits with a soft paper the particles become spheres(150%) and the colour turns to red.

The single - crystal reflectance spectra were recorded from 350to1500 nm with an instrumentation

described in/2/. The absorption spectra were recorded with a Cary 17 spectrophotometer. The Raman spectra were recorded as described previously/3/using Helium-Neon and Argon Lasers and rhodamine dyes. Excitation profiles were measured with respect to $317 {\rm cm}^{-1}$ band of ${\rm CCl}_4$. All measurements were made at room temperature.

The results of measurements are presented in Figs.1,2. Fig.1 shows the single crystal reflectance spectra of $Pt(dapn)_2Pt(dapn)_2Br_2(ClO_4)_4$ in the visible region with the wave vector of light parallel (E || Z) and perpendicular (E || Z) to the needle axis (Z). These spectra show semiconductor behaviour in one dimension(Z). We made a least-square fit to Reflectance (forE || Z) from 350nm to 1500nm using a single Lorentzian dielectric function. The best fit was obtained with the following parameters: ϵ_m =3.40, ω_p =26000cm⁻¹, ω_g =16000cm⁻¹ (625nm), Γ =3300 cm⁻¹. A policrystalline pellet of compound shows qualitatively the same spectrum as that of Fig.1a.

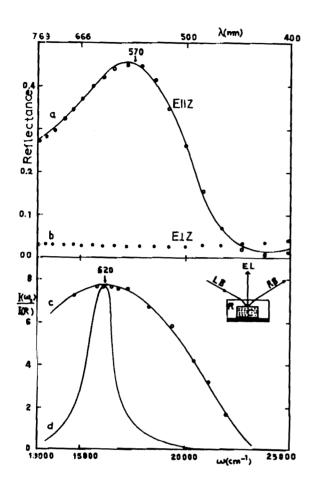
Single crystal Raman spectra of compound show a resonance enhancement of the band $\omega_1 = 179 \text{cm}^{-1}$ due to symmetric vibration of the chain when $E \parallel Z$ and a normal Raman spectrum when $E \perp Z/5/$. Policrystalline pellet shows only the resonance enhancement. The intensity of ω_1 depends strongly on the excit.frequency. Fig.1 shows also the REP obtained from a policrystalline pellet of compound and the calculated one via the Pinczuk-Burstein eqn/6/ (see also/3/). One can see that the maximum occurs near the gap frequency (ω_g) of semiconductor.

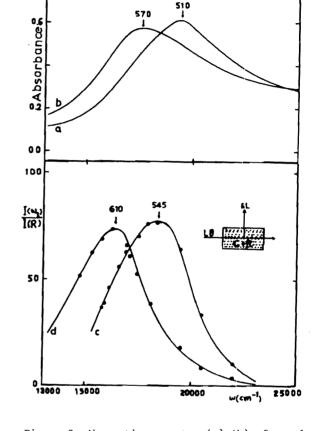
The absorption spectra and REP of suspension in ${\rm CCl}_{\downarrow}$ or of deposits on quartz depend on the size of particles. Figure 2 shows the absorption spectra of samples A and B and the corresponding REP of ${\rm Pt(dapn)}_2{\rm Pt(dapn)}_2{\rm Br}_2({\rm ClO}_4)_4$. One can see that

the maxima are shifted to lower frequencies as the particle size increases. Also we observed that the maximum-positions depends on the shape and the aggregation of the particles. The maxima of needle-like particles and aggregated samples occur at lower frequencies than those of spherical particles

769

666





y(ww)

400

500

Figure 1. Single crystal reflectance spectra with $E\|Z(a)$ and $E\|Z(b)$.ooo experimental data,—Lorentzian fit.REP of a pellet (c) and the calculated one (d). LB=Laser beam, RB=Refl.beam, EL=Entrance lens, R=ref.liquid (CCl_u), C=compound.

Figure 2. Absorption spectra (a),(b) of samples A and B and the corresponding REPs(c),(d).

and non-aggregated samples, respectively. Aggregated samples show REP-maxima near the ω (Fig.1c).In our previous experiments /3/ we did not observe strong shifts in REPs because we used aggregated samples. Also the REPs obtained from the overtone and the anti-Stokes bands showed similar effects.

Similar results have been obtained for other compounds with a metal-halide chain. The size effects in the absorption spectra are attributed to surface plasmons/1/,/7/. The size effects in REP can be considered as a consequence of size quantization of the gap-or exciton-frequency of semiconductor (see/8/-/11/ and refs.therein). The excitonic peaks appear in the spectra of Pt(dapn)₂Pt(dapn)₂Br₂(ClO₄)₄ observed at low temperature/11/.

References

/1/. G.C.Papavassiliou, J.Phys.C:Solid State Phys. 12, L297(1979).

- /2/. G.C.Papavassiliou and A.Zdetsis,J.Chem.Soc:
 Faraday Trans.II,76,104(1980).
- /3/. G.C.Papavassiliou and C.Jacobsen,paper presented in NATO ASI on 1-D Solids,Tomar,Portugal,1979.
- /4/. 0.Bekaroglu et al, Inorg.Chim.Acta <u>21</u>,183 (1977).
- /5/. G.C.Papavassiliou et al, J.Raman Spectr., in press;ibid 8,227(1979).
- /6/. A.Pinczuk and E.Burstein, Proc. 10th Int. Conf. on Phys. of Semiconductors, 1970.
- /7/. G.C.Papavassiliou, "Optical Properties of Small Inorganic and Organic Metal Particles" Prog.Sol.State Chem.12,185(1979).
- /8/. O.N.Filatov and I.A.Karpovich,Sov.Phys.Solid State 10,2284(1969).
- /9/. D.C.Mattis and G.Beni, Phys.Rev.B <u>18</u>,3816(1978). /10/.C.R.Berry, Phys.Rev. <u>161</u>,848(1967).
- /11/.G.C.Papavassiliou et al, to be published.